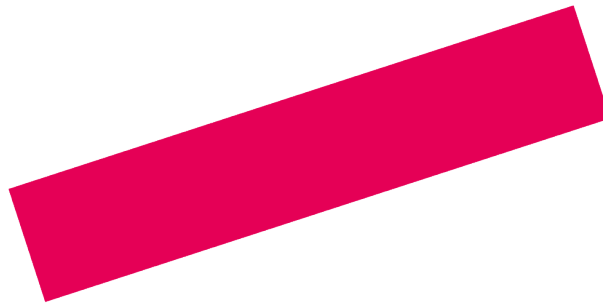


Academie toegepaste biowetenschappen en chemie

# Chemie Course 3



Week 7: H13

# Hoofdstuk 13; alkenen, alkynen en aromatische stoffen

Vandaag

Terugblik op vorige week

Deze week:

- 13.7 Alkeen polymeren
- 13.8 Aromatische stoffen de structuur van benzeen
- 13.9 Benoemen van aromatische stoffen
- 13.10 Reacties van aromatische stoffen

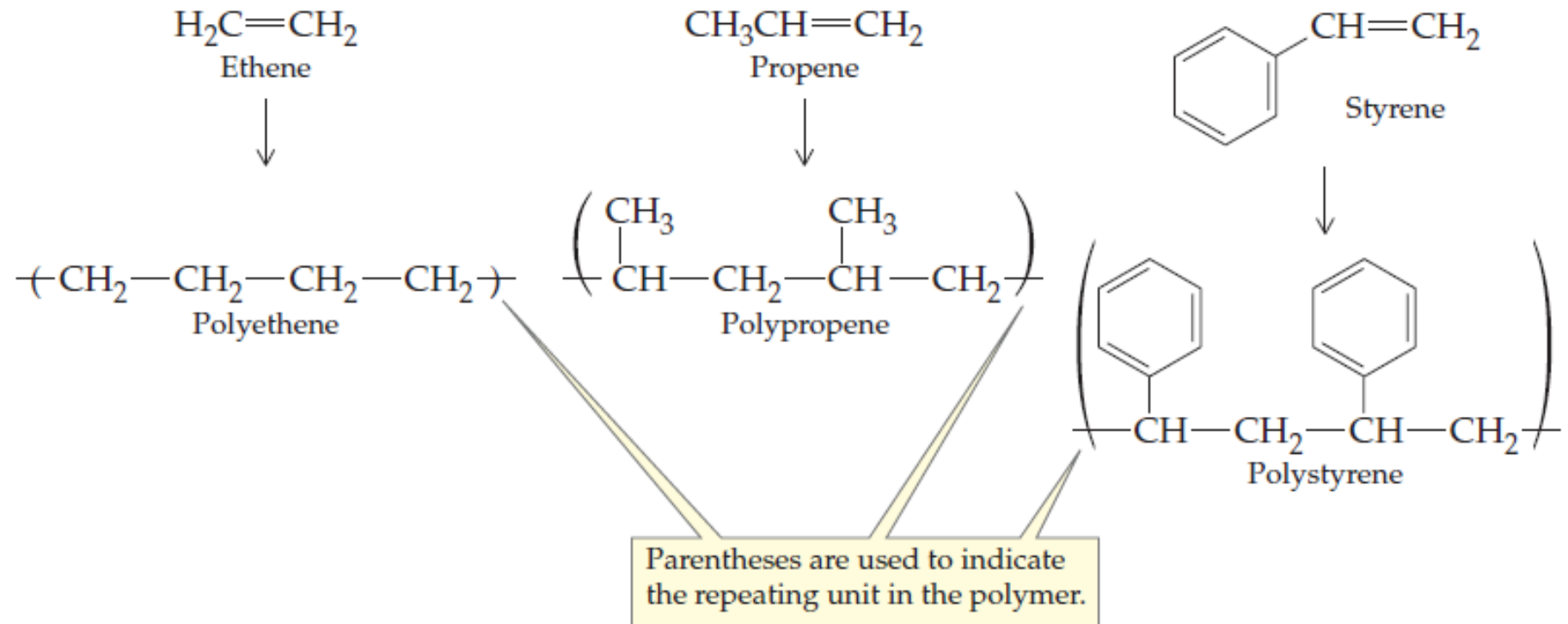


# 13.7 Alkeen polymeren

Wat is een polymeer?

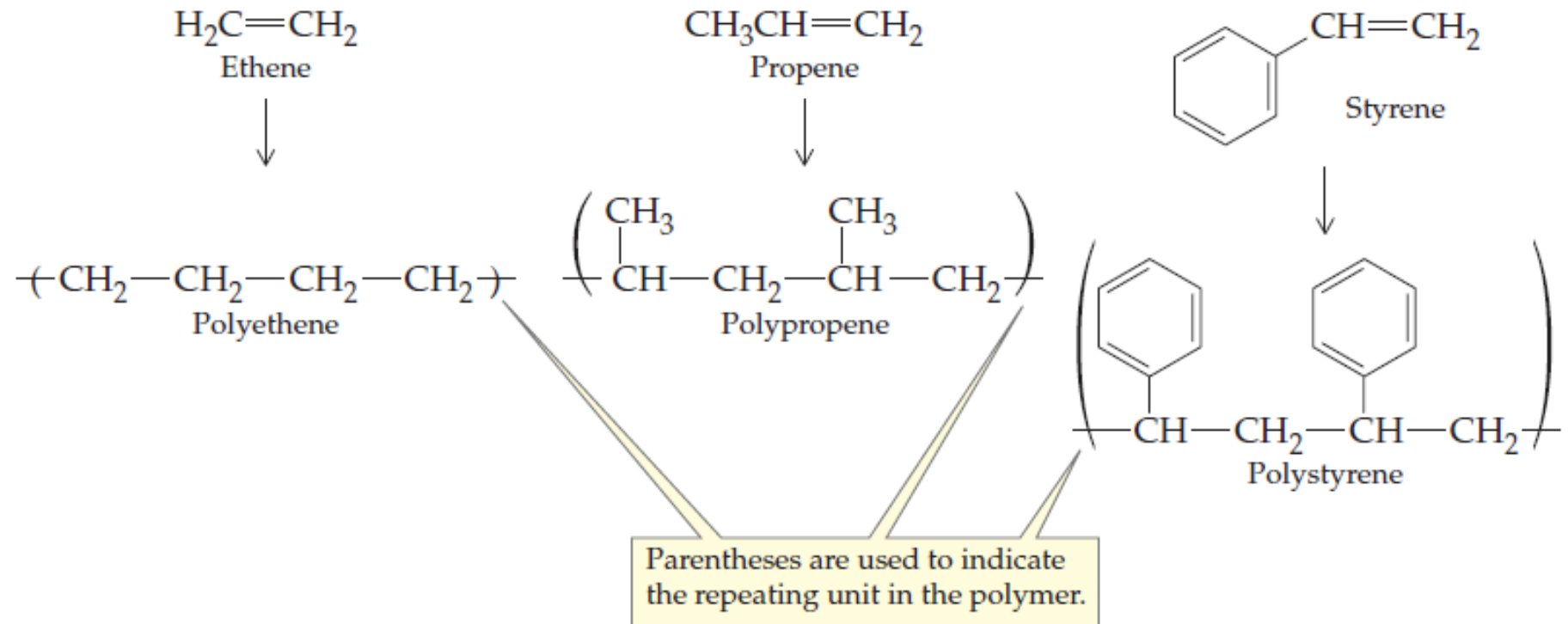
# 13.7 Alkeen polymeren

Een polymeer bestaat uit monomeren.



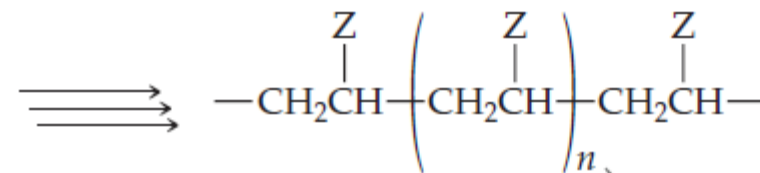
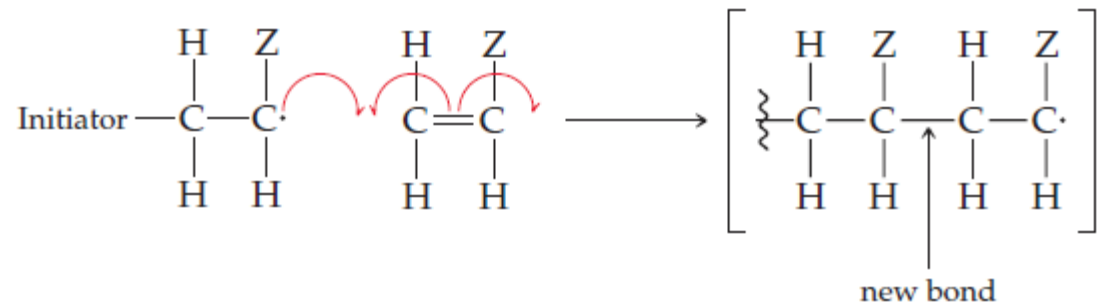
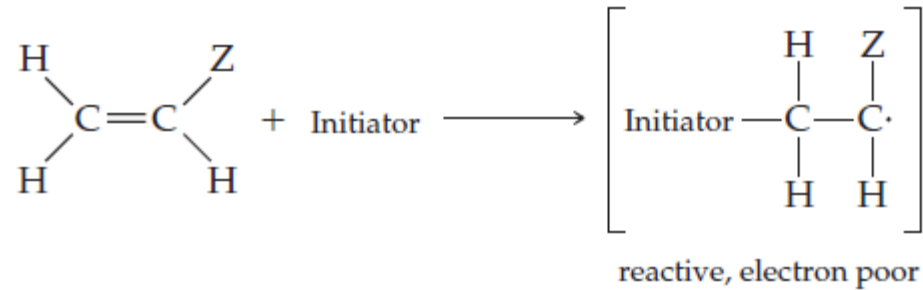
# 13.7 Alkeen polymeren

Een polymeer bestaat uit monomeren.



# 13.7 Alkeen polymeren

Een polymeer bestaat uit monomeren.

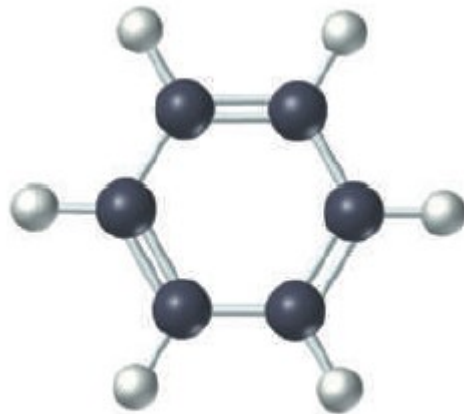


*n* indicates the number of repeating units in the polymer.

# Worked example 13.9

# 13.8 Aromatische stoffen en de structuur van Benzeen

Aromatische stoffen zijn stoffen die een benzeenachtige ring bevatten

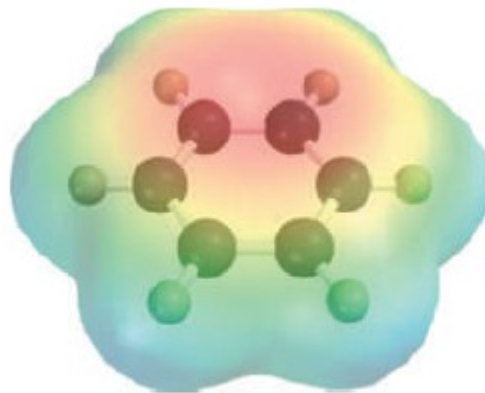
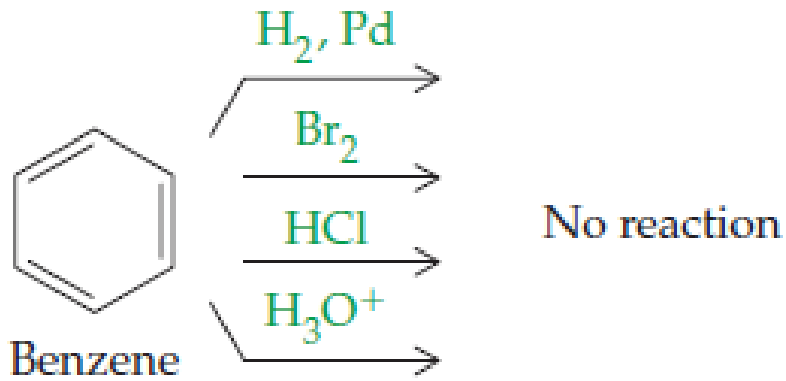


**Aromatic** The class of compounds containing benzene-like rings.

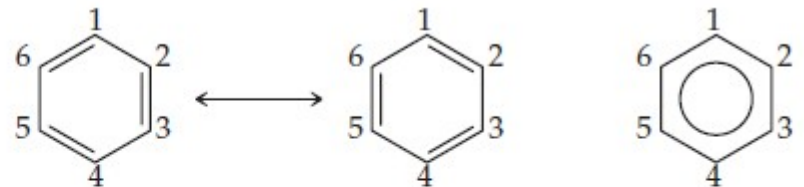


# 13.8 Aromatische stoffen en de structuur van Benzeen

Benzeen is niet reactief



(a)



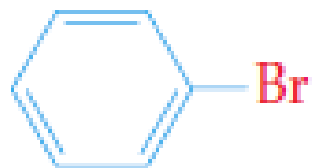
Two equivalent structures, which differ in the position of their double-bond electrons. Neither structure is correct by itself.

(b)

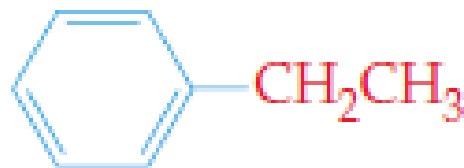
(c)

## 13.9 Benoemen van aromatische stoffen

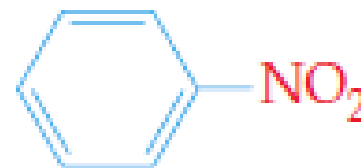
Benzeen is de hoofdnaam in benzenen met een zijgroep.  
bv.



Bromobenzene



Ethylbenzene



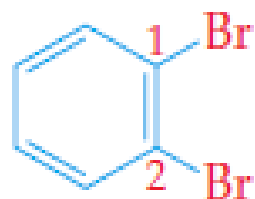
Nitrobenzene

# 13.9 Benoemen van aromatische stoffen

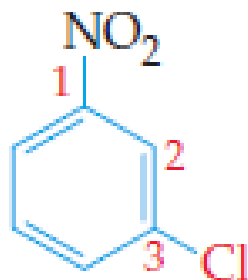
Ortho (1,2)

Meta (1,3)

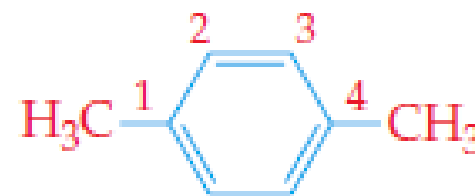
Para (1,4)



1,2-Dibromobenzene  
*ortho*-Dibromobenzene  
*o*-Dibromobenzene



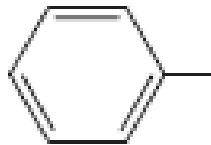
3-Chloronitrobenzene  
*meta*-Chloronitrobenzene  
*m*-Chloronitrobenzene



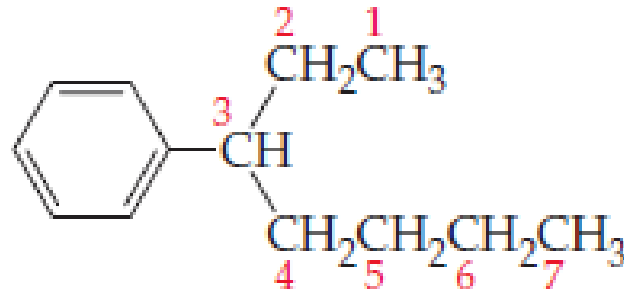
1,4-Dimethylbenzene  
*para*-Dimethylbenzene  
*p*-Dimethylbenzene

# 13.9 Benoemen van aromatische stoffen

De benzeenring kan zelf ook een zijgroep zijn.  
In dat geval wordt het een fenyl genoemd.



A phenyl group  
 $C_6H_5-$

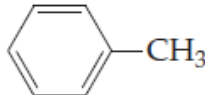
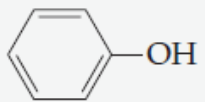
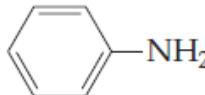

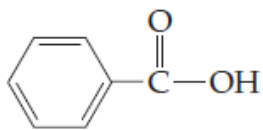
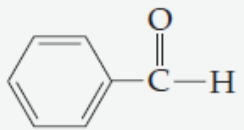


3-Phenylheptane

# 13.9 Benoemen van aromatische stoffen

Wat bekende aromatische groepen.

**Table 13.2** Common Names of Some Aromatic Compounds

Structure	Name
	Toluene
	Phenol
	Aniline
	<i>para</i> -Xylene ( <i>p</i> -Xylene)
	Benzoic acid
	Benzaldehyde

# Worked example 13.10

# Worked example 13.11

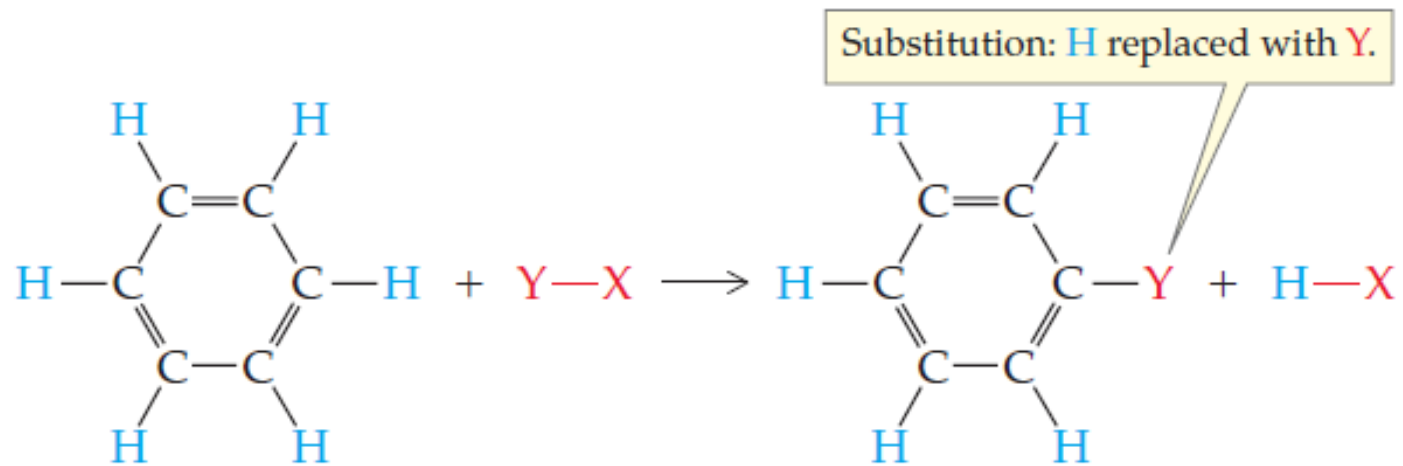
# 13.10 Reactions of aromatic compounds

Aromatische stoffen kunnen substitutie reacties ondergaan.

Wat was ook al weer een substitutiereactie?

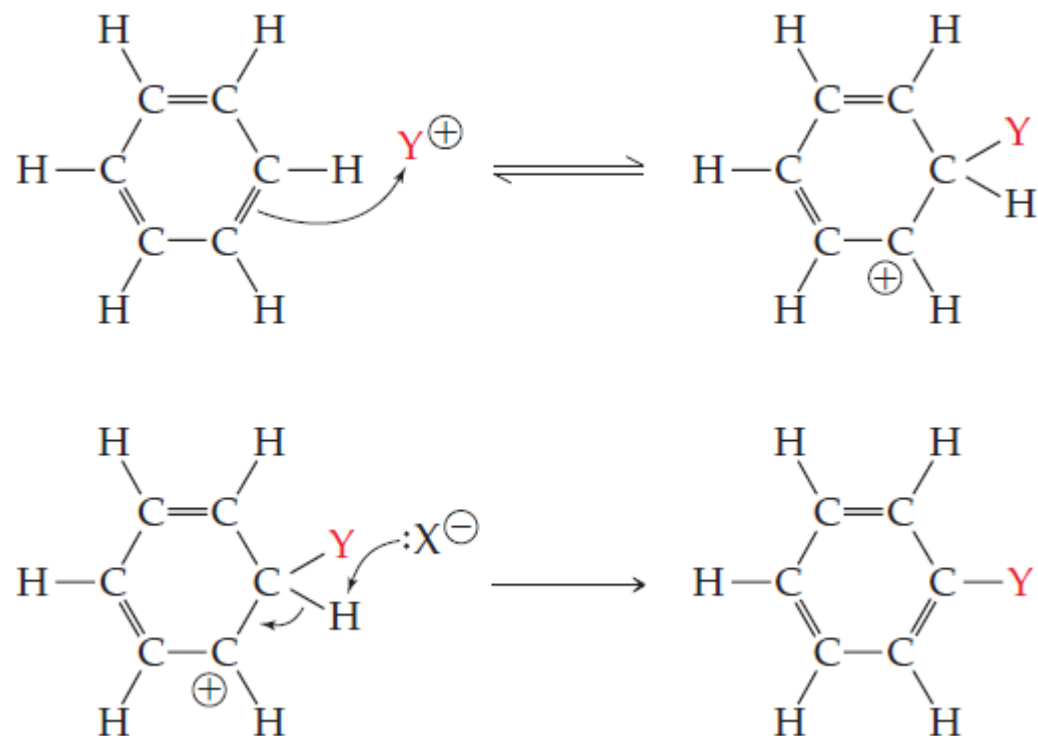
# 13.10 Reactions of aromatic compounds

Aromatische stoffen kunnen substitutie reacties ondergaan.



# 13.10 Reactions of aromatic compounds

De stabiele vorm van de aromatische ring blijft bestaan bij een substitutie.



# 13.10 Reactions of aromatic compounds

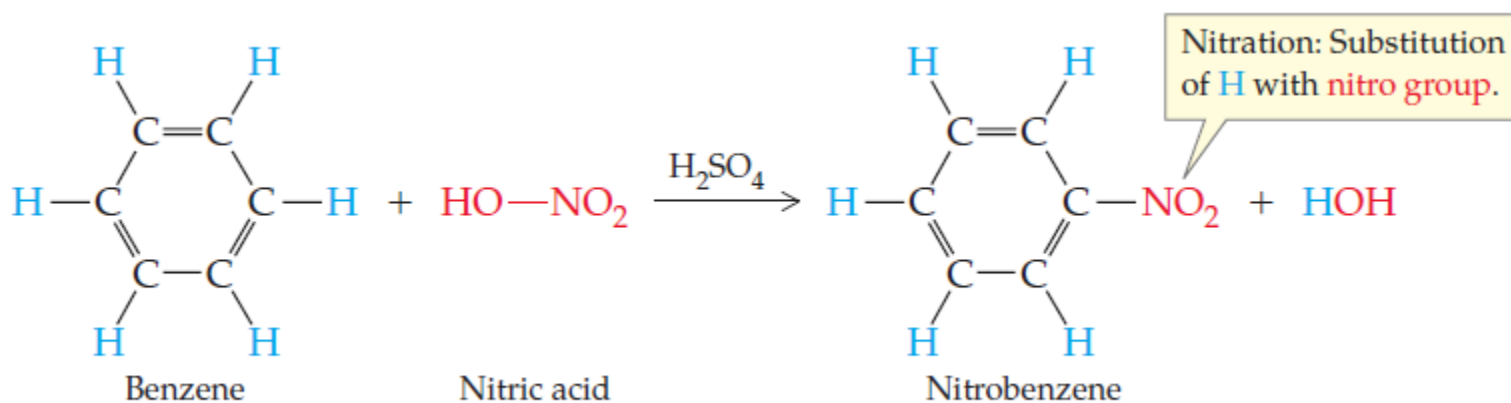
Nitratie

Halogenatie

Sulfonatie

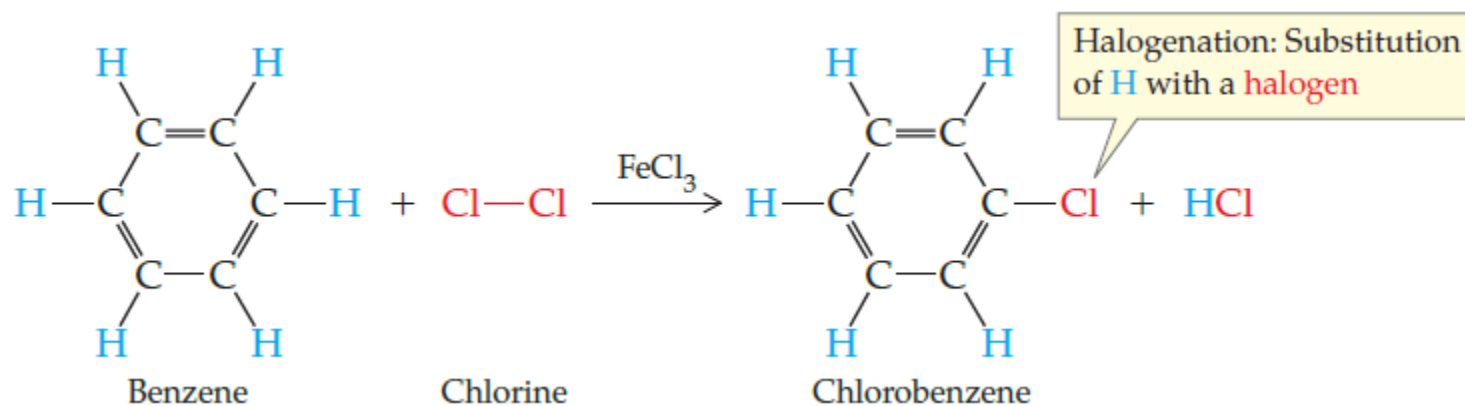
# 13.10 Reactions of aromatic compounds

## Nitration



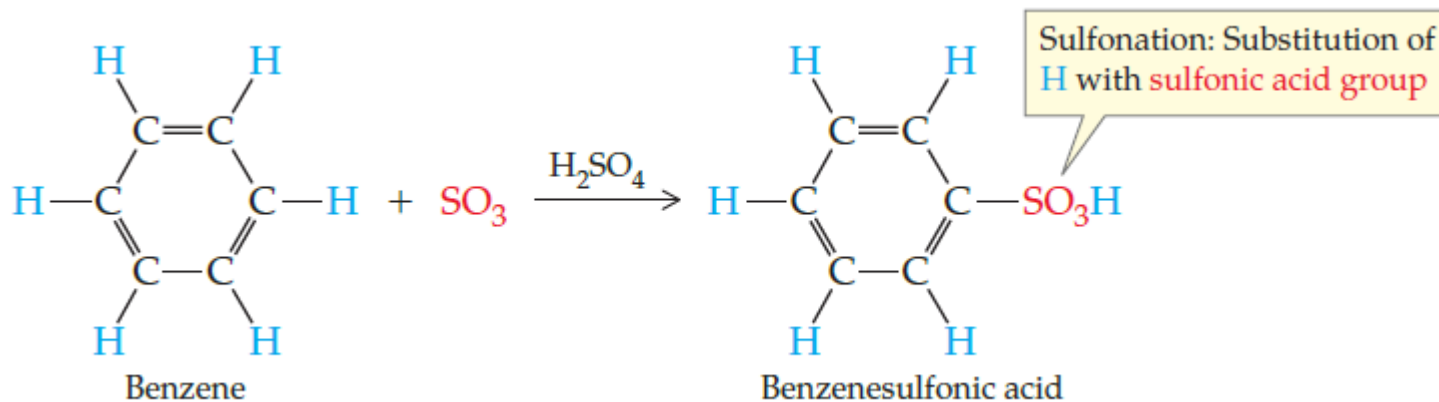
# 13.10 Reactions of aromatic compounds

## Halogenation



# 13.10 Reactions of aromatic compounds

## Sulfonation



# Huiswerk

Maken alle opdrachten H13

t/m 13.23

Zelf nakijken (volgende week vragen stellen)

Oefentoets maken (volgende week nabespreken)

# Bronnen

Afbeeldingen afkomstig van:

- McMurry - Fundamentals of general, organic, and biological chemistry. 7th edition, uitgever: Pearson.
  - Veplicht boek boekenlijst opleiding